

**AVAILABLE AND EMERGING TECHNOLOGIES  
FOR REDUCING GREENHOUSE GAS EMISSIONS  
FROM MUNICIPAL SOLID WASTE LANDFILLS**

# **Available and Emerging Technologies for Reducing Greenhouse Gas Emissions from Municipal Solid Waste Landfills**

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## Abbreviations and Acronyms

ADEME	French Agency for Environmental and Energy Management
ATSDR	Agency for Toxic Substances and Disease Registry
BAAQMD	Bay Area Air Quality Management District
BACT	Best available control technology
Btu	British thermal units
CCAR	California Climate Action Registry
CCTP	Climate Change Technology Program
CEC	California Energy Commission
CH <sub>4</sub>	Methane
CHP	Combined heat and power
CNG	Compressed natural gas
CO	Carbon monoxide
CO <sub>2</sub>	Carbon dioxide
CO <sub>2</sub> e	Carbon dioxide equivalents
CPTR	Cost Incurred Per Metric Ton of Reduced CO <sub>2</sub> e
DER	Distributed Energy Resource
GHG	Greenhouse gas
HAP	Hazard air pollutants
H <sub>2</sub>	Hydrogen
H <sub>2</sub> S	Hydrogen sulfide
kW	Kilowatts
lb	Pound
LFG	Landfill gas
LFGcost	Landfill Gas Energy Cost Model
LFGE	Landfill gas energy
LMOP	Landfill Methane Outreach Program
LNG	Liquefied natural gas
Mg	Megagrams
MSW	Municipal solid waste
MT	Metric ton
MW	Megawatts
MWh	Megawatt-hour
N <sub>2</sub>	Nitrogen
N <sub>2</sub> O	Nitrous oxide
NESHAP	National Emission Standards for Hazardous Air Pollutants
NMOC	Nonmethane organic compounds
NO <sub>x</sub>	Nitrogen oxides
NREL	National Renewable Energy Laboratory
NSPS	New Source Performance Standard
O <sub>2</sub>	Oxygen
ppmv	Parts per million by volume
PSD	Prevention of significant deterioration
psi	Pounds per square inch
RCRA	Resource Conservation and Recovery Act
scfm	Standard cubic feet per minute
SO <sub>x</sub>	Sulfur oxides

SWICS  
WARM

Solid Waste Industry for Climate Solutions  
Waste Reduction Model

## **I. Introduction**

This document is one of several white papers that summarize readily available information on control techniques and measures to mitigate greenhouse gas (GHG) emissions from specific industrial sectors. These white papers are solely intended to provide basic information on GHG control technologies and reduction measures in order to assist States and local air pollution control agencies, tribal authorities, and regulated entities in implementing technologies or measures to reduce GHG under the Clean Air Act, including, where applicable, in permitting under the prevention of significant deterioration (PSD) program and the assessment of best available control technology (BACT). These white papers do not set policy, standards or otherwise establish any binding requirements; such requirements are contained in the applicable EPA regulations and approved state implementation plans.

## **II. Purpose of this Document**

This document provides information on control techniques and measures that are available to mitigate GHG emissions from the municipal solid waste landfill sector at this time. Because the primary GHG emitted by the municipal solid waste landfill industry are methane (CH<sub>4</sub>) and carbon dioxide (CO<sub>2</sub>), the control technologies and measures presented in this document focus on these pollutants. While a large number of available technologies are discussed here, this paper does not necessarily represent all potentially available technologies or measures that that may be considered for any given source for the purposes of reducing its GHG emissions. For example, controls that are applied to other industrial source categories with exhaust streams similar to the municipal solid waste sector may be available through “technology transfer” or new technologies may be developed for use in this sector.

The information presented in this document does not represent U.S. EPA endorsement of any particular control strategy. As such, it should not be construed as EPA approval of a particular control technology or measure, or of the emissions reductions that could be achieved by a particular unit or source under review.

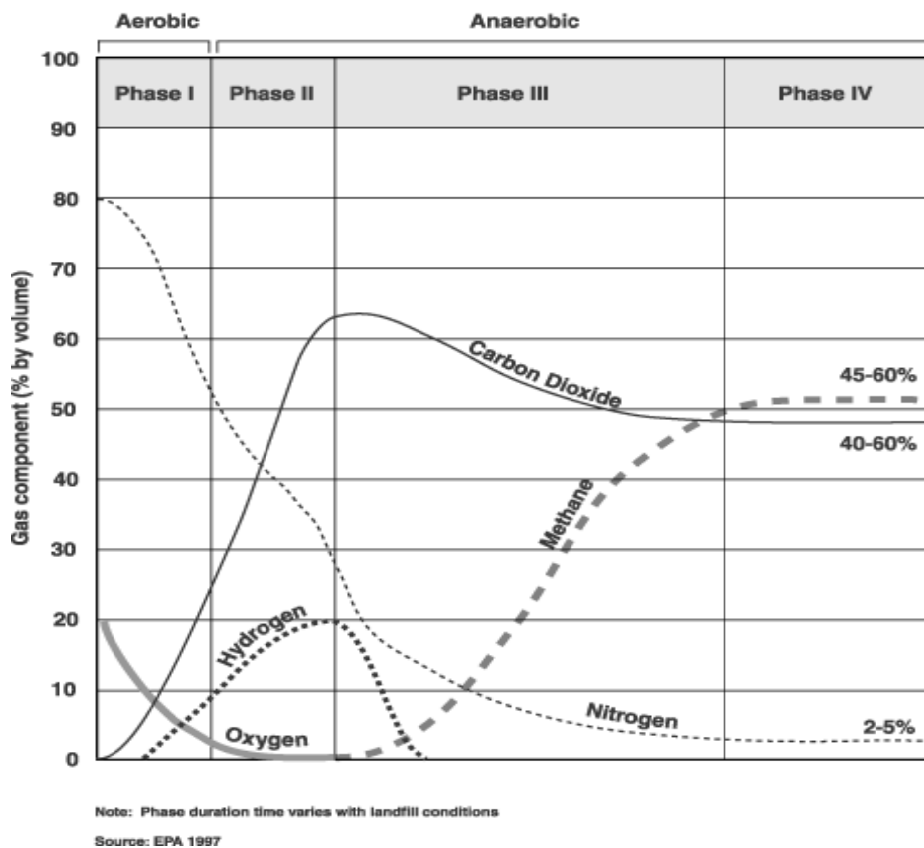
## **III. Description of Municipal Solid Waste Landfills**

The term municipal solid waste (MSW) landfill refers to an entire disposal facility in a contiguous geographic space where municipal waste is placed in or on land. The term does not cover land application units, surface impoundments, injection wells, or waste piles. Many MSW landfills receive other types of waste, such as construction and demolition debris, industrial wastes, and sludge. The information presented in this paper refers to landfills that primarily receive MSW, as defined in the criteria for MSW landfills under the Resource Conservation and Recovery Act (RCRA) regulations (40 CFR Part 258).

According to 2009 data, 54% of MSW in the United States was landfilled, 12% was incinerated, and 34% was recycled or composted (EPA, 2010a). There were approximately

1,800 operational landfills in the United States in 2006 (EPA, 2010b). These landfills accepted approximately 132 million tons of MSW in 2009 (EPA, 2010a).

After placement in a landfill, a portion of organic waste (such as paper, food waste, and yard trimmings) decomposes. Landfill gas is produced by microorganisms under anaerobic conditions and is comprised of approximately 50% CH<sub>4</sub>, 50% CO<sub>2</sub>, and trace amounts of nonmethane organic compounds (NMOC). Landfill gas generation occurs under a four phase process, as shown in Figure 1. First, CO<sub>2</sub> is produced under aerobic conditions. After oxygen (O<sub>2</sub>) is depleted, CO<sub>2</sub> and hydrogen (H<sub>2</sub>) are produced under anaerobic conditions. Then CO<sub>2</sub> production depletes in proportion to the CH<sub>4</sub> that is produced. Finally, CH<sub>4</sub>, CO<sub>2</sub> and nitrogen (N<sub>2</sub>) production stabilize. Significant LFG production typically begins one or two years after waste disposal in a landfill and can continue for 10 to 60 years or longer (ATSDR, 2001a).



**Figure 1: Production phases of landfill gas (ATSDR, 2001a)**

Landfills are the second largest anthropogenic source of CH<sub>4</sub> in the United States; approximately 22% of total U.S. anthropogenic CH<sub>4</sub> in 2008 (EPA, 2010b). The global warming potential of CH<sub>4</sub> is 21 times that of CO<sub>2</sub>, making CH<sub>4</sub> a more potent GHG than CO<sub>2</sub>. Typically, GHG emissions are expressed in terms of carbon dioxide equivalents (CO<sub>2</sub>e) that weigh emissions using global warming potentials. For example, a landfill emitting 1,000 metric tons of

CH<sub>4</sub> and 1,000 metric tons of CO<sub>2</sub> would have CO<sub>2</sub>e emissions of 22,000 metric tons [= (1,000 x 21) + 1,000].

Landfills primarily use the “area fill” method which consists of waste placement on a liner, spreading the waste mass in layers, and compaction with heavy equipment. Daily cover is then applied to the waste mass to prevent odors, blowing litter, scavenging, and vectors (carriers capable of transmitting pathogens from one organism to another). Landfill liners may be comprised of compacted clay or synthetic materials to prevent off-site gas migration and to create an impermeable barrier for leachate. A final cover or cap is placed on top of the landfill, after an area or cell is completed, to prevent erosion, infiltration of precipitation, and for odor and gas control.

Methane generation in landfills is a function of several factors, including: (1) the total amount of waste; (2) the age of the waste, which is related to the amount of waste landfilled annually; (3) the characteristics of the MSW, including the biodegradability of the waste; and (4) the climate where the landfill is located, especially the amount of rainfall. Methane emissions from landfills are a function of methane generation, as discussed above, and (1) the amount of CH<sub>4</sub> that is recovered and either flared or used for energy purposes, and (2) the amount of CH<sub>4</sub> that leaks out of the landfill cover, some of which is oxidized.

#### **IV. Summary of Control Measures**

Table 1 summarizes the GHG control measures presented in this document. Where available, the table includes emission reduction potential, capital costs, operating and maintenance costs, and any important details on the applicability of the control.



**Table 1. Summary of GHG Control Measures for MSW Landfills**

Measure	Applicability	CH <sub>4</sub> Reduction <sup>a</sup>	Typical Capital Costs <sup>b</sup>	Typical Annual O&M Costs <sup>b</sup>	Cost Effectiveness (\$/metric ton of CO <sub>2</sub> e reduced) <sup>e</sup>	Notes/Issues
LFG Collection Efficiency Improvement	All landfills with gas collection systems	Varies	\$24,000/acre	\$4,100/acre	NA	Cost and performance varies depending on the type of cover material.
Flare	All landfills with gas collection systems	99%			\$6 - \$25	Emits secondary criteria pollutant emissions (e.g. NOx and CO).  No revenue.
Turbine	For larger landfills with gas collection systems	99%	\$1,400/kW (≥3 MW)	\$130/kW	\$12 - \$18	Emits secondary criteria pollutant emissions (e.g. NOx and CO).  Generates revenue for landfills.
Engine		96-98%	\$1,700/kW (≥800 kW)	\$180/kW	\$12 - \$16	
Microturbine		99%	\$5,500/kW (≤1 MW)	\$380/kW	\$2 - \$13	
Small Engine		96-98%	\$2,300/kW (≤1 MW)	\$210/kW	\$11	
CHP Engine		96-98%	\$2,300/kW (≥800 kW)	\$180/kW	\$7 - \$57	
CHP Turbine		99%	NA	NA	\$4 - \$51	
CHP Microturbine		99%	NA	NA	\$9 - \$64	
Direct Use (boilers, heaters, etc.)		Varies by technology	\$960/scfm <sup>c</sup> + \$330,000/mile <sup>d</sup>	\$90/scfm <sup>c,d</sup>	NA	
Biocover	All landfills	Up to 32%	\$48,000/acre	NA	\$745	No extensive retrofit.
Biofiltration Bed	Landfills with passive or no gas collection systems	Up to 19%	NA	NA	NA	Low cost.

<sup>a</sup> References provided in section V of this document for the different control measures.

<sup>b</sup> Costs for collection system & flare, turbines, engines, microturbines, small engines, and direct use obtained from *Chapter 4 (Project Economics and Financing)* of LMOP's *Landfill Gas Energy Project Development Handbook* (EPA, 2010c), Costs for CHP engines determined by evaluating the engine case study in the handbook as a CHP engine using LMOP's LFGcost model (EPA, 2010d).

<sup>c</sup> Costs for gas compression and treatment.

<sup>d</sup> Costs for pipeline and condensate management system (if applicable).

<sup>e</sup> Cost effectiveness obtained from analysis done by BAAQMD for conventional landfills with a medium compacted waste density (BAAQMD, 2008), with adjustments made to determine the costs per metric ton of CO<sub>2</sub>e reduced from the combustion of CH<sub>4</sub>, instead of the costs per metric ton of CO<sub>2</sub>e avoided from displacement of power generation. See section V.D and Appendix A for additional information.

NA = not available

## V. Available Control Technologies for GHG Emissions from MSW Landfills

This section describes the available technologies for controlling GHG emissions from MSW landfills. The available control technologies are divided into three categories: LFG collection efficiency improvement, LFG control devices, and increase of CH<sub>4</sub> oxidization. An economic analysis of the control technologies discussed is also included. It should also be noted that large landfills with emissions exceeding 50 megagrams (Mg) NMOC or more are required by New Source Performance Standards (NSPS) to control and/or treat LFG to significantly reduce the amount of toxic air pollutants released. In essentially all cases, controls required by the NSPS will co-control the GHG emissions.

### A. LFG Collection Efficiency Improvement

Collection efficiency is contingent upon landfill design and the manner in which landfills are operated and maintained. Gas collection efficiency can be improved by implementing rigorous gas well and surface monitoring and leak identification and repair. Factors contributing to variability in collection efficiency are discussed below.

There are two types of LFG collection systems, active and passive. Passive systems rely on the natural pressure gradient between the waste mass and the atmosphere to move gas to collection systems. Most passive systems intercept LFG migration and the collected gas is vented to the atmosphere. Active systems use mechanical blowers or compressors to create a vacuum that optimizes LFG collection (ATSDR, 2001a).

For active gas collection systems, the collection efficiency depends primarily upon the design and maintenance of the collection system and the type of materials used to cover the landfill (BAAQMD, 2008). In the background information document for the draft updated landfill AP-42 chapter, a typical collection efficiency range of 50% to 95% is given with a suggested average of 75% (EPA, 2008a).

EPA's Office of Research and Development has completed a field test program using optical remote sensing technology (EPA's OTM-10) to quantify LFG collection efficiency. Sampling was conducted at three MSW landfills to evaluate CH<sub>4</sub> emissions across the landfill footprint to compare to the quantity of extracted gas (i.e., rate of fugitive CH<sub>4</sub> vs. rate of collected CH<sub>4</sub>). The preliminary results suggest gas collection efficiencies from 36% to 85% reflecting a range based on landfill design and operational differences. The report is under review and is expected to be released in 2011.

Higher collection efficiencies may be achieved at landfills with well maintained and operated collection systems, a liner under the waste, and a cover consisting of a geomembrane and a thick layer of clay. Studies conducted by the Solid Waste Industry for Climate Solutions (SWICS) indicate collection systems meeting the requirements of NSPS, Subpart WWW are often more capable of achieving higher collection efficiencies than collection systems used solely for energy recovery because it is difficult to optimize gas quality while trying to attain a high level of gas collection (SWICS, 2009).

Results of gas collection efficiency studies for various cover materials using flux box measurements are documented in Spokas et al. (2005). The data were used to develop default values of percent recovery for the French environment agency (ADEME). These default collection efficiencies for active gas collection systems are listed in Table 2.

**Table 2. LFG Collection Efficiencies for Various Cover Materials**

<b>Type of Landfill Cover Material</b>	<b>Gas Collection Efficiency</b>
Operating cell (no final cover)	35%
Temporary cover	65%
Clay final cover	85%
Geomembrane final cover	90%

Gas collection research studies done by SWICS used flux box data, which may potentially under estimate gas collection efficiency. The resulting collection efficiencies for landfills with active gas collection systems are summarized below (SWICS, 2009):

- 50-70% (mid-range default = 60%) for a landfill or portions of a landfill that are under daily soil cover;
- 54-95% (mid-range default = 75%) for a landfill or portions of a landfill that contain an intermediate soil cover; and
- 90-99% (mid-range default = 95%) for landfills that contain a final soil and/or geomembrane cover systems.

As shown in Table 3, the mid-range default values for the three cover types identified above were adopted as the collection efficiencies listed in the GHG reporting rule for MSW landfills (40 CFR 98, Subpart HH, Table HH-3). The collection efficiency of a passive gas collection system is assumed to be zero because the pressure gradient is unknown and would likely vary in time and space.

**Table 3. LFG Collection Efficiencies in the GHG Reporting Rule**

Description	Gas Collection Efficiency
Area without active gas collection, regardless of cover type	0%
Area with daily soil cover and active gas collection	60%
Area with an intermediate soil cover, or a final soil cover not meeting the criteria below to achieve 95% efficiency, and active gas collection	75%
Area with a final soil cover of 3 feet or thicker of clay and/or geomembrane cover system and active gas collection	95%

As shown in Table 3, landfills with final geomembrane covers have higher collection efficiencies. Changing the final cover material can improve gas collection efficiency. This technology is applicable for all landfills. Typically, modern landfills with active gas collection systems have clay or geomembrane covers in place. An additional geomembrane or clay cover can be added to older landfills with gas collection systems to reduce LFG emissions (BAAQMD, 2008).

### **B. LFG Control Devices**

After collection, LFG may be controlled and/or treated for subsequent sale or use as an energy source to create electricity, steam, heat, or alternate fuels such as pipeline quality gas or vehicle fuel. With approximately half the heating value of natural gas (350 to 600 British thermal units (Btu) per cubic foot), LFG is considered a medium Btu gas. Combustion of LFG is the most common method used to reduce the volatility, global warming potential, and hazards associated with LFG. Combustion methods include destruction devices (e.g., flares), electricity generation units (e.g., reciprocating engines, gas turbines), and energy recovery technologies (e.g., boilers). During the combustion process, CH<sub>4</sub> in LFG is converted to CO<sub>2</sub>. Since CH<sub>4</sub> has 21 times the global warming potential of CO<sub>2</sub>, combustion reduces the global warming effect of LFG significantly. Although CH<sub>4</sub> has 21 times the global warming potential of CO<sub>2</sub>, combusting CH<sub>4</sub> reduces the global warming potential only by a factor of 7.6 because the resulting CO<sub>2</sub> weighs more than the CH<sub>4</sub> by a factor of 2.75. Combustion of LFG also reduces odors and other hazards associated with LFG emissions. However, combustion units emit secondary criteria pollutants, such as carbon monoxide (CO) and nitrogen oxides (NO<sub>x</sub>), as well as hazardous air pollutants (HAP). Fuel cells are considered a non-combustion treatment option for LFG that converts the gas to energy.

The control devices frequently used for LFG and the associated control efficiencies are described in the following sections. It is important to note that all of the technologies discussed

below typically require treatment of the LFG prior to entering the control device to remove moisture, particulates, and other impurities. The level of treatment depends primarily on the type of control and the types and amounts of contaminants in the LFG. A list of common LFG constituents is found in Tables 2.4-1 and 2.4-2 of the landfill AP-42 chapter (EPA, 1998a). Some of the major trace contaminants in LFG that may need to be treated prior to control include sulfur compounds, such as hydrogen sulfide (H<sub>2</sub>S), and siloxanes.

### ***Flares***

Of the combustion methods, flaring is the most commonly used. However, unlike other combustion options, flaring does not recover energy. Controlling LFG emissions by flares is technically feasible for most landfills and many landfills have flares in place. The capital and maintenance costs associated with flares are relatively low compared to other combustion technologies. Flares are often used as backup control devices for landfills that have engines or turbines to generate electricity to limit emissions while these devices are off-line or to respond to variations in LFG generation.

Two different types of flares are available, open flares and enclosed flares. Open flares employ simple technology where the collected gas is combusted in an elevated open burner. A continuous or intermittent pilot light is generally used to maintain the combustion. While open flares are thought to have combustion efficiencies similar to those of enclosed flares, data are not available to confirm this because open-air combustion makes them difficult to test. Under NSPS, Subpart WWW, open flares must meet a minimum Btu content and have a pilot light. For landfills generating LFG that is unable to meet the Btu content consistently, it may be necessary to supplement the collected gas with natural gas or another fuel source, which may create an additional cost for the landfill.

Enclosed flares typically employ multiple burners within fire-resistant walls, which allow them to maintain a relatively constant and limited peak temperature by regulating the supply of combustion air (ATSDR, 2001b). Enclosed flares can be tested for destruction efficiency of NMOC and HAP. The background information document for the draft updated landfill AP-42 chapter provides an NMOC control efficiency range of 86% to 100% for flares, with an average of 97.7% (EPA, 2008a). A report published by California's Bay Area Air Quality Management District (BAAQMD) states that flares typically have CH<sub>4</sub> destruction efficiencies of greater than 99.5% (BAAQMD, 2008). Under NSPS, Subpart WWW, enclosed flares are considered to be incinerators and are required to have a minimum NMOC control efficiency of 98% by weight. In California, flares are required to have minimum CH<sub>4</sub> destruction efficiencies of 99% (CCR, Article 4, Subarticle 6, Section 95464(b)(2)(A)(1)).

### ***Electricity Generation***

Internal combustion engines are the most widely used technology for the conversion of LFG to electricity. Advantages of this technology include: low capital cost, high efficiency, and adaptability to variations in the gas output of landfills. The operation of reciprocating engines at low pressure (12-30 pounds per square inch (psi)) also yields less condensate than operation at

higher pressure (60-160 psi) (Potas, 1993). Internal combustion engines are primarily used at sites where gas production can generate 100 kilowatts (kW) to 3 megawatts (MW) of electricity, or where sustainable LFG flow rates to the engines are approximately 50 to 960 cubic feet per minute (cfm) at 50% CH<sub>4</sub> (EPA, 2010d). For sites able to produce more than 3 MW of electricity, additional engines may be added.

Turbines are an alternative to internal combustion engines. Turbines using LFG require a dependable gas supply for effective operation, and are generally suitable for landfills when gas production can generate at least 3 MW, or where sustainable LFG flow rates to the turbines are over approximately 1,050 cfm at 50% CH<sub>4</sub> (EPA, 2010d). Typically, LFG-fired turbines have capacities greater than 5 MW. Advantages of this technology when compared to internal combustion engines include: a greater resistance to corrosion damage, relatively compact size, and lower operation and maintenance costs. When compared with other generator options, turbines require additional power to run the plant's compression system.

Microturbines can be used instead of internal combustion engines for LFG energy conversion. This technology generally works best for small scale recovery projects that supply electricity to the landfill or to a site that is in close proximity to the landfill. Single microturbine units have capacities ranging between 30 and 250 kW, and are most suitable for applications below 1 MW, or where sustainable LFG flow rates to the microturbines are below approximately 350 cfm at 50% CH<sub>4</sub> (EPA, 2010d). Sufficient LFG treatment is generally required for microturbines and involves the removal of moisture and other contaminants (EPA, 2010c).

In general, turbines have a higher CH<sub>4</sub> destruction efficiency (greater than 99.5%) than internal combustion engines (roughly 96%) (BAAQMD, 2008). For landfills subject to NSPS, Subpart WWW, control technologies are required to have a minimum control efficiency of 98% by weight NMOC reduction or an outlet concentration of 20 parts per million by volume (ppmv), dry basis as hexane at 3% O<sub>2</sub>, of NMOC. In California, LFG control devices other than flares must achieve a CH<sub>4</sub> destruction efficiency of at least 99% by weight; and lean burn internal combustion engines must reduce the outlet CH<sub>4</sub> concentration to less than 3,000 ppmv, dry basis, corrected to 15% O<sub>2</sub> (CCR, Article 4, Subarticle 6, Section 95464(b)(3)(A)). Lean burn internal combustion engines are not defined within this California regulation; however, the NSPS for stationary spark ignition internal combustion engines (40 CFR 60, Subpart JJJJ) defines lean burn engines as any two-stroke or four-stroke spark ignited engine that does not meet the definition of a rich burn engine. Rich burn engines are defined as any four-stroke spark ignited engine where the manufacturer's recommended operating air/fuel ratio divided by the stoichiometric air/fuel ratio at full load conditions is less than or equal to 1.1.

### ***Cogeneration***

Cogeneration, also known as combined heat and power (CHP), is the use of LFG to generate electricity while recovering waste heat from the LFG combustion device. The thermal energy recovered is usually in the form of steam or hot water that can be used for on-site heating, cooling, or process needs. Cogeneration systems are typically more efficient and often more cost effective than separate systems for heat and power (EPA, 2008b). Combustion technologies

generally suitable for CHP include internal combustion engines, gas turbines, and microturbines. There are also boiler/steam turbine applications where LFG is combusted in large boilers for steam generation that is then used by turbines to create electricity (EPA, 2010c).

The CH<sub>4</sub> control efficiency for cogeneration is directly linked to the electricity generation unit combusting LFG. Landfills subject to NSPS, Subpart WWW, must meet the same requirements for cogeneration as those listed above for electricity generation.

### ***Direct Use***

Landfill gas may be used to offset traditional fuel sources such as natural gas, coal, and fuel oil used in industrial, commercial, and institutional applications. Direct use of LFG is primarily limited to facilities within 5 miles of a landfill. There are, however, facilities that have used LFG as a fuel at distances greater than 10 miles. Direct use applications for landfills include: boilers (LFG used solely or co-fired with other fuels), direct thermal technologies (e.g. dryers, heaters, kilns), and leachate evaporation. Innovative uses of LFG include heating greenhouses, firing pottery, glassblowing, metalworking, and heating water for an aquaculture (fish farming) operation (EPA, 2010c).

Control efficiencies of CH<sub>4</sub> for LFG direct use applications vary depending on the type of technology employed. For landfills subject to NSPS, Subpart WWW, control technologies are required to have a minimum control efficiency of 98% by weight NMOC reduction or an outlet concentration of 20 parts per million by volume (ppmv), dry basis as hexane at 3% O<sub>2</sub>, of NMOC. In addition, if a boiler or process heater is used as the control device, the collected LFG must be routed into the flame zone.

### ***Alternate Fuels***

Purification techniques can be used to convert LFG to pipeline-quality natural gas, compressed natural gas (CNG), or liquefied natural gas (LNG). Purification of LFG for the production of natural gas typically involves the removal of inert constituents by adsorption (molecular sieve), absorption with a liquid solvent, and membrane separation. The production of pipeline-quality gas includes processing LFG to increase its energy content and pressurizing the pipeline that is connected to the gas production facility (CCTP, 2005).

The conversion of LFG to CNG and LNG require similar processes, and the resulting products can be used as vehicle fuel. First, the corrosive materials are removed through the use of phase separators, coalescing filters, and activated carbon adsorbents. Next, water and O<sub>2</sub> are removed. A cryogenic purifier is then used to remove CO<sub>2</sub>, which yields high quality gas that is over 90% CH<sub>4</sub> (CCTP, 2005).

The type of LFG alternative fuel production and end use will affect the CH<sub>4</sub> control efficiency. For landfills subject to NSPS, Subpart WWW, control technologies are required to have a minimum control efficiency of 98% by weight NMOC reduction or an outlet concentration of 20 parts per million by volume (ppmv), dry basis as hexane at 3% O<sub>2</sub>, of

NMOC. If the collected gas is routed to a treatment system, including purification and conversion devices, then vented gases from the treatment system must meet these requirements.

### ***Fuel Cells***

A fuel cell is an electrochemical cell that converts energy from a fuel into electrical energy. Electricity is generated from the reaction between a fuel supply and an oxidizing agent. The products of basic fuel cell reactions are CO<sub>2</sub>, water vapor, heat, and electricity (Vargas, 2008). The difference between a battery and a fuel cell is that in a battery, all reactants are present within the battery and are slowly being depleted during the use of the battery. In a fuel cell, reactants (fuel) are continuously supplied to the cell (CEC, 2003). Fuel cells are used in a variety of applications to generate clean electricity without the use of combustion such as in generating transportation fuels for car, boats, and buses. Also fuel cells can serve as a power source in remote locations such as spacecraft, remote weather stations, parks, and in military applications. Fuel cells running on hydrogen are compact and lightweight and have no major moving parts.

For LFG applications, fuel cells use hydrogen from CH<sub>4</sub> to generate electricity (EPA, 1998b). Fuel cells have an advantage over combustion technologies in that the energy efficiency is typically higher without generating combustion by-products such as NO<sub>x</sub>, CO, and sulfur oxides (SO<sub>x</sub>) (EPA, 1998c). If fuel cells are used to generate electricity from landfill CH<sub>4</sub>, then a gas cleanup system is required to ensure that the catalyst within the fuel cell is not contaminated by trace constituents that are present in LFG. Trace constituents include sulfur and chlorine compounds which can inhibit performance and poison the catalyst (NREL, 1998).

EPA's Office of Research and Development conducted a review of fuel cells for LFG applications. The phosphoric acid fuel cell was identified as most appropriate because it is commercially available and has been successfully demonstrated at two landfills. Other types of fuel cells (molten carbonate, solid oxide, polymer electrolyte membrane) may also be applicable for LFG applications as further fuel cell development is conducted. The first demonstration of a fuel cell was at the Penrose Landfill in California. The second was at a Connecticut landfill. Both demonstrations used a 200 kW phosphoric acid fuel cell manufactured by ONSI Corporation (EPA, 1998b). The energy efficiency for the demonstration at the Connecticut landfill was 37% at 120 kW and could have been higher if the waste heat had been utilized. The trace constituents removed in the gas clean up system were flared. An environmental and economic evaluation of a commercial fuel cell energy system concluded that there is a large potential market for fuel cells in this application. The major disadvantage is that the cost is higher compared to combustion technologies such as internal combustion engines and turbines.

For landfills subject to NSPS, Subpart WWW, control technologies are required to have a minimum control efficiency of 98% by weight NMOC reduction or an outlet concentration of 20 parts per million by volume (ppmv), dry basis as hexane at 3% O<sub>2</sub>, of NMOC. If the collected gas is routed to a treatment system, including conversion devices, then vented gases from the treatment system must meet these requirements.



### C. Increase of CH<sub>4</sub> Oxidation

The technologies to increase the CH<sub>4</sub> oxidation rate include biocovers and biofiltration beds. The principle of these technologies is the use of methanotrophic bacteria, which oxidize LFG, specifically CH<sub>4</sub>, to water, CO<sub>2</sub>, and biomass. Methanotrophic bacteria possess the CH<sub>4</sub> mono-oxygenase enzyme that enables them to use CH<sub>4</sub> as a source of energy and as a carbon source. These bacteria are usually found in agricultural soils, forest soils, and compost. These technologies are primarily in the research and development phase, rather than widespread application. The details of these two technologies are discussed below.

#### ***Biocovers***

A biocover is an additional final cover that functions as a CH<sub>4</sub> oxidation enhancer to convert CH<sub>4</sub> into CO<sub>2</sub> prior to venting to the atmosphere. A biocover is composed of two substrate layers: a gas dispersion layer and a CH<sub>4</sub> oxidation layer. The gas dispersion layer is an additional permeable layer of gravel, broken glass, or sand beneath the porous media of the CH<sub>4</sub> metabolizing layer. This layer is added to evenly distribute the uncaptured LFG to the CH<sub>4</sub> oxidation media and to remove excess moisture from the gas. The CH<sub>4</sub> oxidation media can be made of soil, compost, or other porous media. This media is usually seeded with methanotrophic bacteria from the waste decomposition.

This control technology does not require extensive retrofit and is applicable to all landfills, including uncontrolled and older landfills with passive or active collection systems. The biocover itself is not known to affect the functionality of an existing or new gas collection and control system. In addition, it has low secondary criteria pollutant emissions. Biocovers can be used as additional final cover to improve the CH<sub>4</sub> oxidation rate. According to Abichou et al. (2006), biocover applications increased the average CH<sub>4</sub> oxidation rate up to 32%.

#### ***Biofiltration Beds***

Similar to biocovers, biofiltration beds aim to further oxidize CH<sub>4</sub> from passively collected LFG. The collected LFG is passed through a vessel containing CH<sub>4</sub>-oxidizing media prior to venting to the atmosphere or to a control system. This control technology is only feasible for small landfills or landfills with passive gas collection systems due to the size of the biofiltration bed required to treat an air/LFG mixture. In addition, due to the nature of passive gas collection systems, this technology lacks the ability to control and monitor the LFG collection. According to Morales (2006), a pilot project shows that the radial biofiltration bed design has a CH<sub>4</sub> oxidation rate of 19%.

A benefit of using a biofiltration bed compared to LFG combustion is that biofiltration beds produce only CO<sub>2</sub> and water vapor. Unlike other combustion-based mitigation measures, a biofiltration bed does not emit secondary pollutants such as NO<sub>x</sub>, SO<sub>x</sub>, and particulate matter. This technology requires few safety controls for operation, and no start up or shut down procedures.

## D. Economic Analysis

The economic analysis for GHG control technologies is based on a cost effectiveness value, which is defined in this paper as the cost to remove one metric ton of CO<sub>2</sub>e. The cost of LFG control technologies can be estimated using the Landfill Gas Energy Cost Model (LFGcost), which was developed by EPA's Landfill Methane Outreach Program (LMOP) (EPA, 2010d). This model includes direct and indirect costs associated with LFG energy (LFGE) projects. The direct costs are the costs for equipment, including basic treatment of LFG, and installation. The indirect costs include costs for engineering, design, and administration; site surveys and preparation; permits, right-of-ways, and fees; and mobilization/demobilization of construction equipment. Costs estimated by LFGcost are based on costs for average project sites. Individual landfills should adjust costs based on site-specific parameters and conditions. The types of LFG control projects included in LFGcost, Version 2.2 (EPA, 2010d) are as follows:

- LFG collection and flaring systems;
- Direct LFG utilization projects;
- Electricity generation with standard turbines (greater than 3 MW);
- Electricity generation with standard reciprocating engines (800 kW and greater);
- Processing LFG into a high Btu gas (1,000 standard cubic feet per minute (scfm) to 10,000 scfm);
- Electricity generation with microturbines (30 kW to 750 kW);
- Electricity generation with small reciprocating engines (100 kW to 1 MW);
- Leachate evaporators (500 gallons/day and greater);
- Electricity generation and hot water production with CHP reciprocating engines (800 kW and greater);
- Electricity generation and steam production with CHP turbines (greater than 3 MW);
- and
- Electricity generation and steam production with CHP microturbines (30 kW to 300 kW).

In 2008, California's BAAQMD published an economic analysis study on LFG control options using EPA's LFGcost software. This study was performed for MSW landfills of varying sizes (1.5, 3.0, and 5.9 million Mgs), types (conventional and bioreactor), and waste densities (low, medium, and high). The cost effectiveness values contained in the BAAQMD study for electricity generation technologies are based on CO<sub>2</sub>e reduced due to avoided electricity production at the power plant. These values were adjusted to determine cost effectiveness values in terms of CO<sub>2</sub>e reduced from the combustion of CH<sub>4</sub> and CO<sub>2</sub>e reduced from both the combustion of CH<sub>4</sub> and avoided electricity generation. Appendix A contains the calculation procedures used to adjust the original cost effectiveness values in the BAAQMD report. The cost effectiveness for adding LFG combustion options to conventional landfills with a medium compacted waste density (100,000 tons of waste in place per acre) are provided in Table 4.

**Table 4. Cost Effectiveness for Various LFG Combustion Technologies<sup>a, b</sup>**

<b>LFG Combustion Technology</b>	<b>Cost Effectiveness (\$/metric ton of CO<sub>2</sub>e reduced)</b>	<b>Cost Effectiveness (\$/metric ton of CO<sub>2</sub>e reduced and through avoided electricity )</b>
Flare	\$6 - \$25	NA
Engine	\$12 - \$16	\$11 - \$14
Turbine	\$12 - \$18	\$12 - \$16
Microturbine	\$2 - \$13	\$1 - \$12
Small Engine	\$11	\$11
CHP Engine <sup>c</sup>	\$7 - \$57	\$6 - \$52
CHP Turbine <sup>c</sup>	\$4 - \$51	\$4 - \$47
CHP Microturbine <sup>c</sup>	\$9 - \$64	\$8 - \$59

<sup>a</sup> Source: BAAQMD, 2008. Except for flares, values presented in BAAQMD, 2008 were based on CO<sub>2</sub>e avoided through reduction in electricity generated. These values were adjusted to take into account the CO<sub>2</sub>e reduced through combustion of CH<sub>4</sub>. See Appendix A for detailed calculations.

<sup>b</sup> Except for flares, all cost effectiveness values shown do not include costs for the gas collection system. A gas collection system would increase the cost effectiveness by between \$5 and \$10 per metric ton of CO<sub>2</sub>e reduced.

<sup>c</sup> CHP values do not include CO<sub>2</sub>e reductions due to reduction of fuel use where the heat or steam is being used.

In general, it is more economical for larger landfills with high waste densities to install LFG control technologies since their LFG generation rates are higher. The cost of installing combustion technologies is lower for landfills with pre-existing gas collection systems. Flaring is the cheapest combustion technology for most landfills, but flares do not have the potential to generate revenue from LFGE projects.

The cost effectiveness for biocovers was estimated to be \$745 per metric ton of CO<sub>2</sub>e reduced, according to the report prepared by BAAQMD (2008). Since the cost estimates for biocovers were based on a few test sites, the actual cost effectiveness may vary widely.

## VI. Bioreactor Landfill Systems

A bioreactor is typically defined as an MSW landfill where enhanced microbial processes are used to expedite waste decomposition and biological stabilization. To properly manage the stabilization process, certain system design and operational modifications are required (Townsend, 2008). A bioreactor landfill employs the addition of liquid and air into the landfill cell to enhance microbial processes. The most common liquid recirculated in bioreactor landfills is leachate (waste liquid that drains from the landfill), but other liquids may be added to account for lack of moisture in the waste mass (BAAQMD, 2008).

A hybrid (both aerobic and anaerobic enhancements) bioreactor landfill uses two primary processes:

- Air is injected in the top portion of the cell to increase aerobic activity; and
- Liquid is injected into the lower (older) portions of the cell to regulate moisture and promote anaerobic activity (BAAQMD, 2008).

While the term bioreactor is not specifically defined under Subtitle D of the Resource Conservation and Recovery Act (RCRA), there are provisions that allow for short term research, development, and demonstration (RDD) permits specific to bioreactor operations (Townsend, 2008). RCRA Subtitle D prohibits the disposal of bulk liquids unless an RDD permit is granted and allows leachate and LFG condensate recirculation for landfills meeting composite liner requirements. There are also provisions for the prevention of gas migration.

Enhanced degradation in bioreactor landfills also accelerates LFG generation. Compared to conventional landfills, decomposition reaches a higher peak at the year of closure and then declines more rapidly. For anaerobic bioreactors, CH<sub>4</sub> generation rates typically increase 200-250% (Pichtel, 2005). Since LFG is generated more rapidly and the CH<sub>4</sub> concentration in LFG is greater for bioreactor landfills, the gas can be collected and sold for energy recovery earlier than non-bioreactor landfills. To account for accelerated LFG generation and ultimately mitigate GHG emissions from bioreactors, the National Emission Standards for Hazardous Air Pollutants (NESHAP) for landfills requires installation of the collection system and controls prior to liquids addition (40 CFR 63, Subpart AAAA). It should also be noted that under the NESHAP bioreactors are defined as having a minimum average moisture content of 40% by weight. The NESHAP definition of bioreactors also excludes leachate and LFG condensate.

The feasibility of a bioreactor landfill depends on the landfill characteristics and climate. The potential disadvantages of bioreactor landfills include increased odors, physical instability of the waste mass, liner instability, surface seeps, and landfill fires from air addition. Benefits include increased disposal capacity (i.e., more waste can be placed within a fixed volume of landfill air space), shorter post-closure maintenance periods for LFG and leachate management, and better profiles for energy recovery from LFG.

Due to its high capital cost, the implementation of a bioreactor landfill design is suitable primarily for newer active landfill cells that are equipped with the appropriate lining. For existing landfills, converting conventional landfills to bioreactor landfills would require significant changes in landfill design.

## VII. Management Practices

Organic materials account for about 55% of waste currently reaching landfills, primarily consisting of food scraps, yard trimmings, wood, and paper/paperboard (EPA, 2010e). Due to their role as the source of CH<sub>4</sub> in landfills, the diversion of these materials prior to landfilling may be used as a GHG reduction strategy. Diversion methods include composting, recycling, and anaerobic digestion.

Organic waste diversion from landfills prevents CH<sub>4</sub> generation. Methane generation at landfills is reduced proportionally to the amount of organic waste diverted. For example, CH<sub>4</sub> generation at landfills is halved with a 50% organic waste diversion rate. Combining organic waste diversion with a gas collection and control system can further reduce GHG emissions.

Recycling reduces the use of and emissions associated with virgin materials, thus reducing GHG emissions associated with producing the material. Additionally, paper recycling reduces harvesting of trees, thus stabilizing carbon sequestration from forests. According to EPA's Waste Reduction Model (WARM), paper recycling reduces GHG emissions using a lifecycle perspective (EPA, 2010f). There are, however, processing and manufacturing emissions associated with recycling (EPA, 2010e).

Well-managed composting operations facilitate aerobic decomposition. While CH<sub>4</sub> and nitrous oxide (N<sub>2</sub>O) emissions result from anaerobic conditions in the compost pile, a large degree of uncertainty exists in quantifying these emissions. Production of CH<sub>4</sub> and N<sub>2</sub>O from composting varies greatly and results from several factors including: moisture content, carbon-to-nitrogen ratio, stage of the composting process, and the technology used (e.g. windrows, aerated static piles, and in-vessel). While composting operations may reduce GHG emissions, there are emissions associated with pre-processing and on-site equipment (e.g. windrow turners, screens, and blowers); these emissions vary greatly based on the technology used (EPA, 2010e).

Anaerobic digestion is a process where microorganisms break down organic materials in the absence of oxygen. Organic materials are digested in closed containers, minimizing fugitive GHG emissions. Anaerobic digestion yields two products: biogas and a solid residue that can be used as a soil amendment, which can offset conventional fertilizer production and use. Biogas can be used for electricity generation, fuel, or cogeneration.

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## Appendix A

### Calculations to Estimate Cost Effectiveness for CO<sub>2</sub>e Reduced

Cost effectiveness values in Tables 1 and 4 of this paper were derived from cost effectiveness values in a report published by California's Bay Area Air Quality Management District (BAAQMD, 2008). However, the values contained in the BAAQMD report for energy recovery technologies are in units of dollars per metric ton of CO<sub>2</sub>e emissions reduced due to avoided electricity generation at the power plant. The cost effectiveness values from the 2008 BAAQMD report were adjusted to produce cost effectiveness values in units of dollars per metric ton of CO<sub>2</sub>e emissions reduced based on the conversion of CH<sub>4</sub> to CO<sub>2</sub>, a less potent global warming pollutant. Cost effectiveness values were also generated based on the GHG emission reductions from both the conversion of CH<sub>4</sub> to CO<sub>2</sub> (referred to as direct CO<sub>2</sub>e reductions) and the CO<sub>2</sub> emissions avoided from less electricity generated at the power plant (referred to as avoided CO<sub>2</sub>e reductions). This appendix details the calculations for both cost effectiveness values. The cost effectiveness values for flares in the BAAQMD report are based on CH<sub>4</sub> destruction because no energy is recovered (i.e., no electricity avoided); therefore, flare cost effectiveness values were used directly from the report.

The BAAQMD report presents a range of cost effectiveness values for each technology to account for different sized landfills (10 acres, 20 acres & 40 acres). The BAAQMD cost effectiveness values for electricity generation technologies do not include costs for the gas collection system.

#### A.1 Cost Effectiveness Values Based on Direct CO<sub>2</sub>e Reductions

The BAAQMD report referenced the California Climate Action Registry's *General Reporting Protocol* for estimating emission reductions from avoided electricity generation. Tables E.1 and E.3 of the *General Reporting Protocol* contain the 2007 California electricity emission factors listed below. It was assumed that these factors, in units of pounds (lb) per megawatt-hour (MWh), were used to estimate avoided emissions from the power plant (CCAR, 2009).

CO<sub>2</sub> electricity emission rate = 878.71 lb/MWh

CH<sub>4</sub> electricity emission rate = 0.0067 lb/MWh

N<sub>2</sub>O electricity emission rate = 0.0037 lb/MWh

To determine the total amount of CO<sub>2</sub>e reduced from avoided electricity generation, global warming potentials were applied to the CH<sub>4</sub> and N<sub>2</sub>O emission rates. The consolidated CO<sub>2</sub>e emission rate was calculated as follows:

$$\begin{aligned} \text{Overall CO}_2\text{e electricity emission rate} &= (878.71) + (0.0067 \times 21) + (0.0037 \times 310) \\ &= 880 \text{ lb CO}_2\text{e/MWh} \end{aligned}$$

The BAAQMD report utilized LMOP’s LFGcost software (EPA, 2010c). To properly adjust the cost effectiveness values, fuel use rates and efficiencies for each electricity generation technology from LFGcost were used. These default values are provided in Table A-2.

**Table A-2. LFGcost Fuel Use Rates and Efficiencies for LFG Electricity Generation Technologies**

<b>LFG Technology</b>	<b>Fuel Use Rate (Btu/kWh generated)</b>	<b>Efficiency (%)</b>
Engine	11,250	93
Turbine	13,000	88
Microturbine	14,000	83
Small Engine	18,000	92
CHP Engine	11,250	93
CHP Turbine	13,000	88
CHP Microturbine	14,000	83

Source: EPA, 2010c

The example calculation outlined below is for the low value of the cost effectiveness range for engines (\$122 per metric ton (MT) of CO<sub>2</sub>e avoided). The first step is to use the overall CO<sub>2</sub>e electricity emission rate to convert the cost effectiveness value from dollars per metric ton of CO<sub>2</sub>e avoided to dollars per amount of electricity produced (in units of MWh), as follows:

$$(\$122/\text{metric ton CO}_2\text{e}) \times (\text{metric ton}/2205 \text{ lb}) \times (880 \text{ lb CO}_2\text{e}/\text{MWh}) = \$48.7 \text{ per MWh}$$

The second step is to use the appropriate fuel use rate and efficiency from Table 2 and the heat content and density of CH<sub>4</sub> to calculate the cost in terms of dollars per metric ton of CH<sub>4</sub> produced by the landfill, as follows:

$$(\$48.7/\text{MWh}) \times (\text{MWh}/1000 \text{ kWh}) \times (\text{kWh}/11,250 \text{ Btu}) \times 0.93 \times (1012 \text{ Btu}/\text{ft}^3 \text{ CH}_4) \times (\text{ft}^3 \text{ CH}_4/0.0423 \text{ lb CH}_4) \times (2205 \text{ lb}/\text{metric ton}) = \$212 \text{ per metric ton CH}_4$$

The next step is to calculate the amount of CO<sub>2</sub>e reduced from the conversion of CH<sub>4</sub> to CO<sub>2</sub>. The global warming potential of CH<sub>4</sub> is 21, which is used to express the amount of CH<sub>4</sub> destroyed in terms of CO<sub>2</sub>e. The amount of CO<sub>2</sub> generated from the combustion of CH<sub>4</sub> must be subtracted from the amount of CH<sub>4</sub> destroyed using a mass balance method to result in an accurate measure of CO<sub>2</sub>e reduced. The overall CO<sub>2</sub>e reduced is calculated as:

$$\begin{aligned} \text{CO}_2\text{e reduced} &= (\text{CH}_4 \text{ destroyed as CO}_2\text{e}) - (\text{CO}_2 \text{ generated by CH}_4 \text{ combustion}) \\ \text{CO}_2\text{e reduced} &= (21 \text{ metric tons CO}_2\text{e/metric ton CH}_4) - (44 \text{ metric tons CO}_2/16 \text{ metric tons CH}_4) \\ \text{CO}_2\text{e reduced} &= 18.25 \text{ metric tons CO}_2\text{e per metric ton CH}_4 \end{aligned}$$

Lastly, the dollars per metric ton of CH<sub>4</sub> produced by the landfill are divided by the overall CO<sub>2</sub>e reduced to estimate the cost effectiveness values in terms of dollars per metric ton of direct CO<sub>2</sub>e reduced, as follows:

$$\begin{aligned} \text{Adjusted cost effectiveness} &= (\$212/\text{metric ton CH}_4) \times (\text{metric ton CH}_4/18.25 \text{ metric tons CO}_2\text{e}) \\ \text{Adjusted cost effectiveness} &= \$12 \text{ per metric ton direct CO}_2\text{e reduced} \end{aligned}$$

Tables 1 and 4 in the main section of this paper contain the adjusted cost effectiveness values for direct CO<sub>2</sub>e reduced for all seven electricity generation technologies.

## A.2 Cost Effectiveness Values Based on Direct and Avoided CO<sub>2</sub>e Reductions

The original cost effectiveness values in the 2008 BAAQMD report represent avoided CO<sub>2</sub>e reductions and the adjusted cost effectiveness values, as discussed in section A.1, represent direct CO<sub>2</sub>e reductions. Therefore, the calculation of cost effectiveness values that represent both direct and avoided CO<sub>2</sub>e reductions can be accomplished using the original and adjusted cost effectiveness values. The derivation of the equation used to determine cost effectiveness values in units of dollars per metric ton of direct and avoided CO<sub>2</sub>e reductions is as follows:

$$\begin{aligned} \$/D &= \$ \text{ per metric ton of direct CO}_2\text{e reduced} \\ \$/A &= \$ \text{ per metric ton of avoided CO}_2\text{e reduced} \\ \$(D+A) &= \$ \text{ per metric ton of direct and avoided CO}_2\text{e reduced} \end{aligned}$$

$$\begin{aligned} \$(D+A) &= (\$/A) / ((D+A)/A) = (\$/A) / ((D/A) + (A/A)) = (\$/A) / ((D/A) + 1) \\ \$(D+A) &= (\$/A) / ((\$/A)/(\$/D) + 1) \end{aligned}$$

Using the example calculation for the low cost effectiveness value for engines from section A.1, the cost effectiveness value for direct and avoided CO<sub>2</sub>e reduced is calculated as:

$$\begin{aligned} \$/D &= \$12/\text{metric ton of direct CO}_2\text{e reduced} \\ \$/A &= \$122/\text{metric ton of avoided CO}_2\text{e reduced} \end{aligned}$$

$$\$(D+A) = (\$122/\text{metric ton}) / ((\$122/\text{metric ton})/(\$12/\text{metric ton}) + 1) = \$11/\text{metric ton}$$

$$\text{Cost effectiveness for direct \& avoided CO}_2\text{e reduced} = \$11 \text{ per metric ton of CO}_2\text{e reduced}$$

Table 4 in the main section of this paper contains the cost effectiveness values for direct and avoided CO<sub>2</sub>e reduced for all seven electricity generation technologies.